


Heavy Metal Contamination in Soil and Vegetables: A Review with Health Risk Assessments



Sha Md. Shahan Shahriar*, Md. Sarwar Hossain, Sumaiya Dipti, and Sayed M A Salam

Department of Applied Chemistry and Chemical Engineering, University of Rajshahi, Rajshahi-6205, Bangladesh.

*Corresponding author: e-mail address: shahan@ru.ac.bd; ORCID: <https://orcid.org/0000-0003-3956-2409>

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ARTICLE INFO

Received: 06th November, 2023

Revised: 04th December, 2023

Accepted: 15th December, 2023

Keywords:

contamination
food web
health risk
heavy metal
soil
vegetable

ABSTRACT

Since the dawn of time, plant-based foodstuffs have been an indispensable component of human nutrition. This hasn't evolved, but the heavy metal contamination of soil and edible vegetable portions is currently a global grave threat to the environment. One of the predominant components contributing to agricultural contamination includes heavy metals. Anthropogenic activities and rapid industrialization can introduce dangerous and invisible heavy metals throughout the soil, water, air, and plants, among other environmental components. In addition to being critical for plants to flourish consistently, heavy metals also play substantial functions in basic nucleic acid metabolisms, electron transfer, redox reactions, and as direct participants in several enzymes. It is crucial that these vital metals be present in growth media at a certain concentration, yet an excess of them might have detrimental consequences ranging from deadly ailments. The article reviewed the existing understanding of how those released toxic heavy metals penetrate the food chain, biomagnify into cells when they are consumed as vegetables, and cause potentially catastrophic consequences to health. These harmful metals have a significantly higher propensity to bioaccumulate and turn deadly in human beings.

1. Introduction

Over the past 200 years, urbanization and industrialization have grown at a rapid rate, resulting in significant levels of air, water, and soil pollution. Because of their toxicity, persistence, and capacity to build up in the biota, trace metals in the environment continue to pose a hazard even if persistent organic contaminants and greenhouse gases have received more attention [1]. Cadmium (Cd), Chromium (Cr), mercury (Hg), arsenic (As), lead (Pb), and other metalloids having a density of greater than 5 g/cm³ are examples of heavy metals. Besides polluting the environment, heavy metals can seriously injure humans by building up in bodily organs and other living things via the food chain [2, 3]. Exposure to heavy metals in soil primarily occurs via inhalation, oral consumption, and skin touch, constituting the three main routes of human contact [4, 5]. The main causes of heavy-metal pollution include milling, industrialization,

burning fossil fuels, agrochemicals, and mining which release a range of HHMs into agricultural soils and water bodies, including Pb, Cr, Co, Cd, Zn, As, Ni, Cu and Hg [6-8]. Because they include essential nutrients that humans need to survive, vegetables are a vital part of a typical diet. Additionally, they serve as protective meals by helping people prevent disorders. Vegetables may gather greater concentrations of hazardous heavy metal when cultivated near heavy metal pollution sources [9]. The accumulation of heavy metals in vegetables as well as crops that are edible in contaminated soils eventually poses a major risk to the human body as well as animal health as a result of the detrimental and irreversible effects of metals [10]. Few heavy metals such as Co [11], Mn [12], Zn [13], Cu [14] and Ni [15] are key components at certain levels in humans yet they become toxic when exposed to higher doses. Heavy metals pose significant health risks, particularly Pb along with Cd, which can cause damage for example malformations and

bone fractures, along with hypertension, kidney failure, cardiovascular problems and other severe nervous system, immune system, lung and liver diseases [16-18]. Memory loss, heart problems, digestive disorders, and genotoxicity are among the negative consequences of mercury poisoning [19]. Many analytical methods and instruments use mercury (Hg), such as barometers and thermometers, blood pressure monitors with fluorescent illumination, amalgam for dental restoration, and fluorescent lighting [20]. Heavy metal accumulation at chronic level in the liver, kidney and bones of humans can result in kidney, cardiovascular, neurological, and bone illnesses. This buildup can be caused by long-term ingestion of high quantities of heavy metals through contaminated food. These days, as more people become aware of the health dangers linked with heavy metal exposure, risk assessment has become a global hot subject [21].

2. Heavy metal sources causing soil and vegetable contamination

Plant food contamination can result from air pollution or by coming into touch with polluted soil. Heavy metals from various sources can deteriorate the soil, which is essential to the growth of food crops. Anthropogenic activities related to agriculture and industry are the main causes of soil pollution with heavy metals worldwide [22-24]. Soil-crop ecosystems are impacted by sewage from industries, irrigation, agricultural practices, and mining activities in highly populated and developing nations [25]. The use of specific inorganic and organic fertilisers, insecticides, herbicides, and organic manure that contain heavy metals can potentially boost heavy metal content in soils [26, 27]. The swift advancement of industrialization around the globe has led to a notable rise in the likelihood of heavy metal pollution in the environment. Toxic materials build up in the soil, air, water as a consequence of rapid industrialization, chaotic urbanisation, and long-term heavy fertiliser and pesticide usage [28, 29]. In part because of the higher concentrations of heavy metals and metalloids in leachates from municipal solid waste dumps, inadequate garbage disposal methods can also affect soil-crop systems [30]. The amount of heavy metals in food is directly impacted by soil contamination since samples of plant food taken from contaminated soils have been shown to accumulate these hazardous elements. [31-33]. Different origins of those heavy metals and their injurious proliferation on vegetables as well as crops are given in **Table 1**.

3. Heavy metal contamination in soil

Heavy metal accumulation in soil is the most significant. Heavy metals, which include chromium (Cr), arsenic (As), copper (Cu), cadmium (Cd), mercury (Hg), zinc (Zn), nickel (Ni), and lead (Pb) are often found contaminants in soil environments. This kind of pollution is pervasive, long-lasting, and physiologically hazardous in the soil

environment [44]. A vital component of any ecosystem is soil. Because of its capacity to both absorb and emit, soil may get contaminated from various sources [45]. The weathering of rocks, volcanic activity, and erosion are the primary natural sources of the incorporation of heavy metals into soils. Anthropogenic sources include smelting, ore mining operations, landfills, industry effluents, electroplating, military training and warfare, use of phosphate fertiliser and pesticides, biosolids utilisation (such as fertilisers, animal manure, and municipal effluent), and atmospheric deposition [46, 47]. The excess soil contamination scale is 16.1%, with the highest over-standard concentrations of heavy metals being Pb, Cd, Hg, As, and Cr at 1.50%, 7.0%, 1.60%, 2.70%, and 1.10%, respectively [48]. The overall state of the soil in agricultural settings can serve as a marker for heavy metal pollution brought on by farming practices [49]. The soil has already begun to absorb the majority of these heavy metals. On the contrary, long-term exposure to heavy metals can cause bone fractures and lung cancer in humans [50, 51]. Remediation methods for heavy metal-polluted soil have been extensively studied. These methods include ex-situ remediation methods (soil washing, landfilling, vitrification, solidification) and in-situ methods (electrokinetic process of extraction, encapsulation, surface capping, chemical immobilisation, phytoremediation, soil flushing, and bioremediation) [52-54].

4. Mechanism of toxic heavy metals accumulation in vegetables

Notwithstanding the mass amount of heavy metals in aerial tissues and excluders, different plants as well as vegetables serve as accumulators and evidence of survival. Plants that successfully undergo biodegradation or biotransformation of pollutants into innocuous arrangements inside their tissues have higher rates of survival [62]. Plants accumulate hazardous heavy metals through various mechanisms, including absorption by root, xylem loading, heavy metal mobilisation, root-to-shoot transit, cellular compartmentation, and sequestration. The bioaccumulations of heavy metals in different organs of vegetables are given in **Table 2**. Almost all heavy metals are found in soil in an insoluble form which is not bioavailable to plants. Different root exudates that plants release can change the rhizosphere's pH along with enhancing the solubility of heavy metals, hence increasing their bioavailability [63, 64].

Vegetables absorb the heavy metals found in soil organic matter, and when they connect to short carbon atom chains, they transform into crucial cations [65]. In root hairs, heavy metals are actively transferred to the root cell plasma membrane from the apoplast or the soil. Upward movement, vacuoles, and barriers in the xylem as well as phloem are crucial for the translocation of the absorbed metals to the shoot [66]. The bioavailable metal penetrates the cell wall to access the root cells after being absorbed at the root surface. The two main pathways by which heavy metals are taken by

roots are the symplastic pathway (active transport against electrochemical potential gradients and concentration across the plasma membrane) and the apoplastic pathway (passive diffusion) [67].

this is insufficient to initiate the methylation of As in plants [68]. As is catalyzed by specific enzymes, which includes Sadenosylmethionine-dependent arsenite methyltransferase, which is also crucial for the biogeochemical pathway of As

Table 1. Different origins of heavy metals and their toxic contamination of vegetable /crops

Source of Heavy Metals	Heavy Metal Present	Contamination	Vegetable/crop	Region	Reference
Industries	Cd, Co, Cr, Cu, Fe, Mg, Mn, Pb, Zn	Leafy vegetables	<i>Lactuca sativa</i> , <i>Raphanus sativus</i> , <i>Mentha piperita</i> , <i>Petroselinum crispum</i> , <i>Eruca Sativa</i> , <i>Daucus carota</i> , <i>Trigonella foenum-graecum</i> , <i>Allium sativum</i> , <i>Zingiber officinale</i> , <i>Solanum tuberosum</i> , <i>Raphanus sativus</i> , <i>Allium cepa</i> etc	Jazan, Saudi Arabia	[34]
Soil	Pb, Hg, Cd, As	Vegetable and fruit	<i>Beta vulgaris L</i> , <i>Raphanus sativus L</i> , <i>Daucus carota</i> , <i>Brassica rapa</i> , <i>Brassica nigra</i> , <i>Brassica oleracea</i> , <i>Spinacia oleracea L.</i> , <i>Coriandrum sativum</i> , <i>Mentha</i>	Maharashtra, India	[35]
Soil	Cr, Cd, Pb, Ni, Cu, Zn, Fe, Mn	Leafy and root vegetables	<i>Brassica oleracea</i> , <i>Solanum lycopersicum</i>	Dhaka, Bangladesh	[36]
Contaminated waterbody	As, Zn, Cd, Pb, Cu, Hg, Co	Vegetables and soil	<i>Solanum tuberosum</i> , <i>Cucumis sativus</i> , <i>Solanum lycopersicum</i> , <i>Daucus carota</i> , <i>Lactuca sativa</i> , <i>Spinacia oleracea</i>	Koka area of central Ethiopia	[37]
Soil	Cu, Ni, Fe, Zn, Cd, Pb, As, Mn, N, P	Vegetable	<i>Avena sativa L.</i> , <i>Cynara scolymus L.</i> , <i>Medicago sativa L.</i> , <i>Corchorus olitorius L.</i> , <i>Cynodon dactylon L.</i>	Isfahan, Iran	[38]
Wastewater	Cd, Cu, Zn, Al, Cr, Co, Mn, Fe, Pb, V, Mo	Soils and vegetables	<i>Triticum turgidum</i> , <i>Eruca. Sativa</i> , <i>Malus sylvestris</i> , <i>Vicia faba</i> , <i>Triticum aestivum</i> , <i>Madia sativa</i> and <i>Urtica dioica</i>	Asmara, Eritrea	[39]
Wastewater treatment plant	Pb, Cd, Zn, Cu	Water, soil, crops	<i>Cucurbita moschata</i> , <i>Spinacia oleracea</i> , <i>Amaranthus lividus</i> , <i>Basella alba</i> , and <i>Trichosanthes cucumerina</i>	Marrakech, Morocco	[31]
Wastewater of Shitalakhya River	Cu, Ni, Cd, Cr, Pb, Zn	Vegetables, Soil	<i>Phaseolus vulgaris</i> , <i>Raphanus raphanistrum</i> , <i>Brassica rapa</i> , <i>Brassica oleracea</i> , <i>Spinacia oleracea</i> , <i>Solanum melongena</i> , and <i>Piper nigrum</i>	Narayangonj, Bangladesh	[40]
Soil	Hg	Vegetables	<i>Piper nigrum</i> , <i>Amaranthus dubius</i> , <i>Phaseolus vulgaris</i> , <i>Solanum melongena</i> , <i>Ipomoea batatas</i> , <i>Solanum lycopersicum</i> , and <i>Ipomoea aquatic</i>	China	[41]
Region Impacted by Mine	Pb, As, Cd, Cu	Vegetables, Soil	<i>Oryza sativa</i>	Daye, China	[42]
Commercial and municipality sewage water	Ni, Zn, Pb, Cd	Crops, Soil		Lenjan, Iran	[43]

soil pH, dissolved cations and anions, chemical pesticides, and species of vegetation, fertilisers, and soil type all affect how much hazardous metal is taken up from the soil [62]. In the rhizosphere, bacteria generate DMA (dimethyl arsenic acid) and MMA (monomethyl arsenic acid), nevertheless, <https://doi.org/10.62275/josep.24.1000007>

[69]. Numerous metal transporters are involved in the mobilisation along with the transfer of metals from one compartment of cells and plant organs to another, including OsHMA2, OsHMA3, OsNramp5, Lsi1, NIP, Lsi2, MMA, HAC, ABC, DMA, and PT [70].

Heavy metal ions have a tendency to combine with a variety of chelators, including organic acids, after they have entered root cells. These complexes, which include phosphate precipitate, sulphate, and carbonate, are subsequently immobilised in either the intracellular (symplastic compartments, including vacuoles) or extracellular (apoplastic cellular walls) [71]. Through the root symplasm, the metal ions contained inside the vacuoles may travel into the stele as well as the xylem stream [72]. Heavy metals are mostly accumulated in leaves by xylem-loading absorption, and in seeds and fruits through phloem-loading [73] are then transported by xylem vessels to the shoots. They are carried and dispersed throughout leaves by apoplast or symplast, where they are confined in extracellular compartments (cell walls) or plant vacuoles, preventing the buildup of free metal ions in the cytosol [74]. Metal ion transporters and complexing agents are two of the many molecules that mediate the uptake and translocation of heavy metals in plants. The absorption of heavy metal ions from soil depends on these specialised transporters, channel proteins, also known as H⁺-coupled carrier proteins, which are present in the plasma membrane of root cells. In addition to mediating the influx-efflux of metal translocation from roots to shoots, they are capable of moving certain metals across cellular membranes [75]. Several heavy metal ions, for example, Fe²⁺, Co²⁺, Cd²⁺, Mn²⁺, and Cu²⁺ are also transported by the naturally resistant associated macrophage proteins (NRAMPs) [76, 77]. In addition to metal ion transporters, amino acids as well as organic acids, which are complexing agents, function as metal ligands to contribute to the chelation of heavy metal ions. For instance, in the xylem, citrate also is a significant chelator for Ni and Fe [67, 71, 78]. The bioaccumulations of heavy metals in different organs of vegetables are presented in Table 2.

5. Health risk assessment

A human health risk assessment (HRA) is the method of determining the kind along probability of adverse health outcomes in individuals who could be subjected to chemicals or other potentially dangerous substances in the environment. An HRA is a type of health inquiry designed to evaluate a person's health risks and quality of life [16]. This section describes some of the parameters found in the literature used to calculate the HRA.

5.1 Bio-concentration factor (BCF)

BCF is an essential term in health risk assessment since it provides quantifiable information on a contaminant's capacity to be taken up by organisms. It is frequently employed as a preliminary screening measure for bioaccumulative, hazardous, and persistent chemicals. The BCF of heavy metals in soils and vegetables was calculated by dividing the concentration of each heavy metal in the edible parts of the vegetables by the concentration of the heavy metal in the soil sample. A lower BCF score indicates less heavy metal

transport from soil to crops. BCF values larger than one, on the other hand, indicate that the investigated plants absorb more heavy metals from the soil [79]. The bioconcentration factor is assessed by following Eq. 1

$$BCf = \frac{Cv}{Cs} \quad (1)$$

Where, Cs = heavy metals in soil samples (ppm), Cv = heavy metals in vegetables on a dry weight (DW) basis (ppm),

5.2 Estimated daily intake (EDI)

Daily intake is determined by both metal content in food and daily food consumption. Furthermore, the human body weight can impact contamination tolerance. The EDI is a notion that was developed to account for these concerns. Eq. 2 is used to calculate the anticipated daily intake for each element. The EDI value of each metal may be used to calculate the health risk of a vegetable.

$$EDI = \frac{E_f \times E_D \times F_{IR} \times C_M \times C_f}{B_W \times T_A} \times 10^{-3} \quad (2)$$

Where E_f denotes exposure frequency (365 days/year); E_D means exposure duration, equivalent to average lifetime; F_{IR} is equal to average food (vegetable) consumption; C_M denotes metal concentration (ppm dry weight); C_f is equal to 0.085 which is concentration conversion factor; B_w denotes reference body weight for an adult; T_A means the average exposure time in 65 years (23,725 days) and 10⁻³ is unit conversion factor [80-82].

5.3 Target hazard quotient (THQ)

The target hazard quotient (THQ) is the ratio of hazardous element exposure to the reference dosage, which is the greatest amount at which no adverse health consequences are predicted. The THQ method is used to assess the non-carcinogenic danger of eating infected crops [83].

$$THQ = \frac{EDI}{RfD} \quad (3)$$

Where RfD is the oral reference dose (mg/kg/day). If THQ is less than 1, the exposed population is unlikely to suffer visibly adverse repercussions. When THQ is equal to or more than 1, there is a potential health concern, and proper preventative measures and steps should be taken.

5.4 The carcinogenic risk index (CRI)

The carcinogenic risk index determines a human's lifetime risk of acquiring cancer. CRI is used to assess the potential human health risk of recognised carcinogens. The CRI is computed by multiplying the EDI by the oral cancer slope factor (CSF) for the heavy metals in question [84]. The following formula is used to calculate the CRI.

$$CRI = EDI \times CSF \quad (4)$$

When the CRI is larger than, the exposed people are at a major carcinogenic risk; but, when the CRI is less than, the

exposed populations are not at a significant carcinogenic risk. Furthermore, if TR and/or TCR levels are between 10^{-4} and 10^{-6} , the exposed populations are at high risk of cancer [84].

another exciting field that can remediate heavy metal-containing wastewater before it is released into the environment. By using these technologies, heavy metal

Table 2. Bioaccumulation of heavy metals in different organs of vegetables

Scientific name of the vegetables	Organ of accumulation	Heavy metal concentration, mg/kg	Location	Reference
Peri-urban vegetable	Edible parts	Cd- 0.30; Hg- 0.26; As- 0.37; Pb- 0.54; Cr- 1.17; Cu- 6.17; Ni- 1.96; Zn- 18.56	China	[55]
<i>Coriandrum sativum</i> , <i>Allium cepa</i> , <i>Solanum lycopersicum</i>	Edible parts	Cd- 0.13-0.23; Pb- 0.46-2.12; Ni- 0.54-0.89; Co- 0.22-0.47; Zn- 16.77-36.65; Cu- 14.46-21.65	Jhansi, India	[56]
<i>Lepidium sativum</i> , <i>Coriandrum sativum</i> , <i>Spinacia oleracea</i>	Leaves	Pb- 0.2-4.7; Zn- 1.7-6.7; Cd- 0.02-0.52; Cu- 0.59-2.23; Ni- 0.04-0.81; As- 0.018-0.126; Co- 0.005-0.21; Cr- 0.1-1.87	Tehran, Iran	[57]
<i>Spinacia oleracea</i> , <i>Solanum melongena</i> , <i>Cucumis melo</i>	Edible parts	Mn- 18.7-137; Cu- 22.2-65; Ni- 1.8-5.05; Zn- 19.5-41; Cd- 0.05-0.39; Fe- 129-968; Cr- 2.7-3.7; Pb- BDL	Pakistan	[58]
<i>Lactuca sativa</i> , <i>Raphanus sativus</i> , <i>Mentha piperita</i> , <i>Petroselinum crispum</i> , <i>Eruca Sativa</i>	edible portions of leafy vegetable	Cd- 0.242–1.249; Co- 0.457–0.764; Cr- 1.10–3.534; Cu- 5.36-9.18; Fe- 355.4-1195.9; Mn- 31.90-132.63; Pb- 0.858-1.175; Zn- 18.20-36.72	Saudi Arabia	[34]
<i>Brassica oleracea</i> , <i>Beta vulgaris</i> , <i>Lactuca sativa</i> , <i>Solanum lycopersicum</i>	Edible parts	Pb- 0.061-0.741; Cr- 0.136-1.141; Cd- 0.028-0.477; Zn- 0.012-8.72	Ethiopia	[59]
<i>Amaranthus hybridus</i> , <i>Ipomoea batatas</i> , <i>Abelmoschus esculentus</i> , <i>Solanum melongena</i>	Edible parts	Zn- 2.64-10.29, Fe- 4.84-13.64 Pb- 0.32-2.46; Cu - 0.55-1.04	Tanzania	[60]
<i>Solanum lycopersicum</i> , <i>Solanum cepa</i> , <i>Capsicum annum</i>	Edible parts	Pb- 0.02-0.03; Ni- 0.02-0.48; Cr- 0.001-0.02; Fe- 0.74-4.37; Cd- 0.001-0.004; Zn- 0.50-1.16; Cu- 0.14-0.25	Turkey	[61]

5. Conclusion:

Food supply chain contamination is one of the noteworthy approaches that heavy metals penetrate human body. It is potentially dangerous to human beings and impedes the worldwide supply of food. The main causes of food chain contamination are inadequate effluent management, lax enforcement of standards and laws, a lack of research evidence on sustainable management, and low public consciousness. In the end, food safety is concealed by these constraints. This review has uncovered the root causes of heavy metal pollution, how they contaminate food chains, how plants carry metals, and the way our bodies absorb them. Transportation of heavy metals into plants, including fruits and vegetables, is mostly due to soil pollution. For this reason, attaining food safety and security on a global scale is hampered by the buildup of heavy metal(loid)s in agricultural soils. Although bioremediation is a potential natural remedy for heavy metal(loid) pollution, several problems require to be resolved before it can be used more widely. Accelerating global soil mapping and developing regional models capable of accurately predicting pollutant distributions and pinpointing contamination sources will be advantageous. Biosorbents offer appealing prospects as an inexpensive way to prevent pollution in the environment. Nanotechnology is

content in the soil may be decreased, and as a result, the bioaccumulation of heavy metals in vegetables can be reduced as well. Heavy metal removal technologies comprise electrolytic recovery, membrane filtration, ion exchange, adsorption, precipitation, and more. It is indispensable that further research be conducted to develop more pocket friendly, sophisticated, and ecologically conscious versions of them to fulfil the prerequisites for the removal of heavy metals from our soil as well as food web.

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